

Strengthen Students' Character through Inquiry Based Thermochemical I Experiment

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Abstract—The 21st century skills needed is a quality of character, and basic literacy in order to realize the gold generation in 2045. Strengthening of the characters on all levels of education is to strengthen moral values, morality, and the personality of students by strengthening the character education integrated in all subjects. This study aims to describe the reinforcement of student character through thermochemical I experiment through inquiry based worksheet. The subjects were 20 students. Data were collected using observation sheets and student worksheets based on inquiry. Research results showed strengthening of the character of students through inquiry-based thermochemical I experiment performed in nine stages there were habituation to read verses from the Al Qur'an, implementation of the pre-test, pre laboratory, observation, manipulation or formulating problems, make hypotheses, verification, data analysis and make conclusions. Based on the steps taken, the expected strengthening of the characters are religious, love reading, honest, discipline, responsibility, curiosity, tolerance, hard work, friendly and appreciate achievement.

Keywords—based Inquiry; experiment; strengthening of character; thermochemistry I

I. INTRODUCTION

Efforts to strengthen character is higher than moral formation, because character formation is not only concerned with what is right or wrong, but how to instill habits about things that are good in life. Thus, learners have high awareness and understanding, as well as concern and commitment to implement the virtues in everyday life [1]. Pattern of character strengthening in each institution is different. This is due to different rules. Higher education as a culmination point of the formal education process has a role and responsibility in strengthening the character of the nation that has been built and developed since primary education.

Strengthening the character through the experiment is very likely to realize the character values [2]. Inquiry learning through the experiment in laboratory can instill student attitudes [3]. This is in line with one of the national educational goals that is faithful and cautious to God Almighty, have noble character, and responsible" [4].

Research on character development has been done by [5] that character can be formed through exemplary and habituation through learning process. In addition character can be formed through the learning process associated with daily life [6]. Character can be developed from a scientific-based practice [7]. In contrast with the researches, researchers put more emphasis on strengthening the character through inquiry based-worksheet on Thermochemical practice I.

II. RESEARCH METHODS

Descriptive research method was used to describe the process of conducting thermochemical I practice that is identification of instant heater and cooler pouch in reinforcing the character of students [8]. The sampling technique is using one group of practice which amounted to 20 students. The instrument used is an observation sheet that contains character enhancement that can be developed through the thermochemical I experiment and the inquiry-based student worksheet that contains the practice procedure on instant heater and cooler pouch. The instrument is processed by means of descriptions of characters that appear in each stage of inquiry, and then take percentage of occurrence of the characters on a scale of 100.

Stages of research conducted on the basis of the stages of inquiry study were observation, manipulation, hypothesis, verification, data analysis and make conclusions [9]. The research procedure covered every student starts practice by first reading Al-Quran for 10 minutes, followed by pre-test on student ability in predict sample that can be used as material of instant heater and cooler, pre-test was done for 10 minutes. The lecturer gave pre-laboratory on the evaluation of preliminary journal that had been made by the students and the procedure of experiment to be performed as well as the character control. The experiment was done by using the Thermochemistry I worksheet on an inquiry-based instant heater and cooler pouch. After experiment the student returned the tools and materials that have been used. Then complete the post-laboratory followed by a discussion of the results of the experiment performed and prepared the practice report. Scheduled picket group have to clean laboratory. From the beginning to completion of experiment activities the character of students observed by five laboratory assistants.

The students' experimental processes are as follows: each sample is weighed two grams with analytic balance. The volume of aquades is measured as 15 mL by using a measuring cup. Then measure the initial temperature of the aquades with the thermometer. Then input the aquades and calcium chloride into the calorimeter, then stir, observe and record the temperature change every 30 seconds for five minutes. Do it for samples of ammonium nitrate and calcium carbonate. The characters expected from the stage of verification are **responsibility, honest, tolerance, discipline, hard work and friendly.**

Stage of data analysis done based on experiment procedure, so obtained by observation data based on table 1 below:

TABLE I. OBSERVATIONAL DATA

No	material	Treatment	Observation
1	Samples A Calcium Chloride	Calcium chloride is weighed	Two grams
		The volume of aquades is measured	A total of 15 ml
		The initial temperature of the aquades is measured	The initial temperature of 28 ° C
		Aquades are reacted with calcium chloride in the calorimeter	An increase in temperature from 28°C to 32°C is done per 30 seconds for 5 minutes
		Changes that occurred were observed	Chloride diets are soluble in water and include exothermic reactions
2	Sample B Ammonium Nitrate	Ammonium Nitrate is weighed	Two grams
		The volume of aquades is measured	A total of 15 ml
		The initial temperature of the aquades is measured	The initial temperature of 28 ° C
		Aquades are reacted with ammonium nitrate in the calorimeter	A temperature drop from 28° C to 26° C is performed per 30 seconds for 5 minutes
		Changes that occurred were observed	Ammonium nitrate is soluble in water and includes an endothermic reaction
3	Sample C Calcium Carbonate	Calcium carbonate is weighed	Two grams
		The volume of aquades is measured	A total of 15 ml
		The initial temperature of the aquades is measured	The initial temperature of 28 ° C
		Aquades are reacted with Calcium carbonate in the calorimeter	An increase in temperature from 28°C to 29°C is done per 30 seconds for 5 minutes
		Changes that occurred were observed	Occurs sediment and color of white solution excluding endothermic or exothermic reactions

Graph of observation data of temperature change of sample A (calcium chloride) can be seen in Figure 3 below:

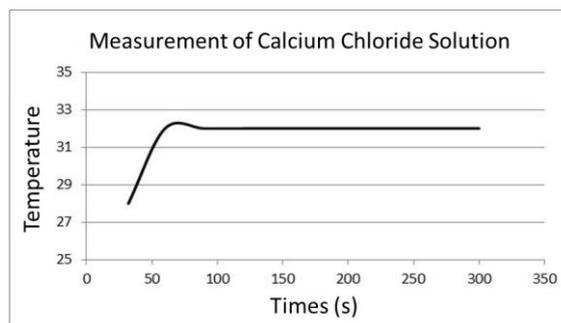


Fig. 3. Graph of calcium chloride temperature changes.

Based on the figure of 3 reactions between calcium chloride and water temperature rise, from the initial temperature of 28°C to 32°C. This reaction includes exothermic reaction, the reaction that occurs when the release of heat. The calcium chloride is suitable for use as an instant heater pouch.

Graph of observation data of sample B that is Ammonium Nitrate can be seen in figure 4 below:

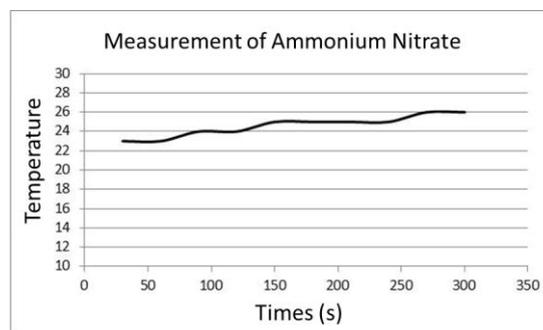


Fig. 4. Graph ammonium nitrate temperature changes.

Based on Figure 4 the reaction between ammonium nitrate with water to experience deterioration in temperature, from the initial temperature of 28°C to 26 °C include endothermic reaction, the reaction that occurs when the ongoing heat absorption. Thus ammonium nitrate suitable for use as the instant cooler pouch.

Graph of observation data of sample C that is calcium carbonate can be seen in figure 5 below:

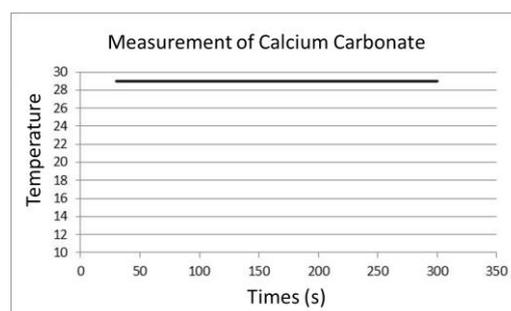


Fig. 5. Graph of changes in calcium carbonate temperature.

Based on the figure of 5 reactions between calcium carbonate and water, the reaction will run very strongly and quickly when in powder form, calcium carbonate powder will release the heat. The molecule of CaCO_3 will soon bind the water molecule (H_2O) which will form calcium hydroxide, a soft substance such as a paste. As indicated in the reaction as follows: $\text{CaCO}_3 + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2 + \text{CO}_2$. Thus, calcium carbonate is not suitable as the instant cooler or heater pouch. Based on the data analysis stage, expected characters **are honest, appreciate achievement and like to read.**

The stages to make conclusions are based on data analysis. The conclusions made by the students based on the experiments carried out are materials suitable for use as instant heater filler is calcium chloride due to exothermic reaction and as the instant cooler filler is ammonium nitrate due to endothermic reaction.

IV. CONCLUSION

Strengthening the character of students through the inquiry-based thermochemical I practice is done in nine stages of reading the scripture Al-Qur'an, Implementation of pre-test, pre-laboratory, observation stage, manipulation or formulating the problem, making hypotheses, verification, data analysis and make conclusions. Based on the steps taken, the expected strengthening of the character is religious, reading, honest,

discipline, responsibility, curiosity, tolerance, hard work, friendly and appreciate achievement.

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